## Preparation of Macrodiolides via a Common Chiral Building Block. Total Synthesis of (-)-Pyrenophorin and (-)-Pyrenophorol

Nobuo Machinaga and Chihiro Kibayashi\* Tokyo College of Pharmacy, Horinouchi, Hachioji, Tokyo 192-03, Japan

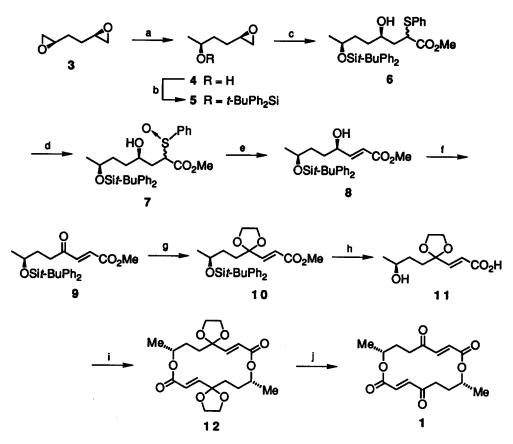
Abstract: Macrodiolides (-)-pyrenophorin and (-)-pyrenophorol have been synthesized utilizing a  $C_2$  symmetric (R,R)-diepoxide as a common enantiopure chiral building block.

The macrolide dilactone (diolide) (-)-pyrenophorin  $(1)^1$  is an antifungal antibiotic produced by the plant pathogenic fungi *Pyrenophora avenae* and *Stemphylium radicinum*, which is closely related structurally to (-)pyrenophorol (2) isolated from *Byssochlamys nivea*<sup>2a</sup> and *Stemphylium radicinum*.<sup>2b</sup> These naturally occurring diolides are characterized by a 16-membered ring derived by head-to-tail dimerization of two identical C<sub>8</sub> hydroxy acid subunits. Although intensive efforts have been directed toward the synthesis of 1,<sup>3</sup> almost all synthetic approaches reported are based upon the preparation of the racemic monomeric hydroxy acids followed by dimerization to the diolide, which leads to the formation of 1:1 mixture of the racemic and meso forms of 1.<sup>3,4</sup> Hitherto, only a few examples of chiral synthesis of the naturally occurring 8*R*,16*R* isomer 1 have been published by Seebach's group<sup>5</sup> and Takano et al.<sup>6</sup> In contrast to the comprehensive work done on 1, it was only very recently that the first synthesis of pyrenopholrol (2) was accomplished,<sup>7</sup> which enabled the relative and absolute natural configurations of 2 to be established.



Recent investigation from this laboratory has revealed that the chiral  $C_2$  symmetric diepoxide 3 serves as a versatile building block in the preparation of several natural products.<sup>8</sup> In this communication we now report a enantioselective synthesis of these natural diolides 1 and 2 based on a stereo-defined approach utilizing the enantiopure (*R*,*R*)-diepoxide 3 as a single, common chiral building block.

As shown in Scheme 1, the synthesis of 1 was undertaken to obtain the monoepoxide 4 by monoaddition of hydride ion to the diepoxide  $3^9$  which was best effected by using 1 mol equiv of Vitride in THF, leading to the epoxy alcohol 4 in 74% yield based on 40% recovered starting material. After O-silylation (*t*-BuPh<sub>2</sub>SiCl, DMAP), the resulting monoepoxide 5 was further subjected to ring opening by exposure to (phenylthio)acetic acid dianion (PhSCH<sub>2</sub>CO<sub>2</sub>H, LDA, THF) followed by diazomethane esterification, affording the  $\alpha$ -phenylthio- $\gamma$ -hydroxyester 6 in 48% overall yield from 4. The sulfoxide 7, which resulted from sodium metaperiodate Scheme 1



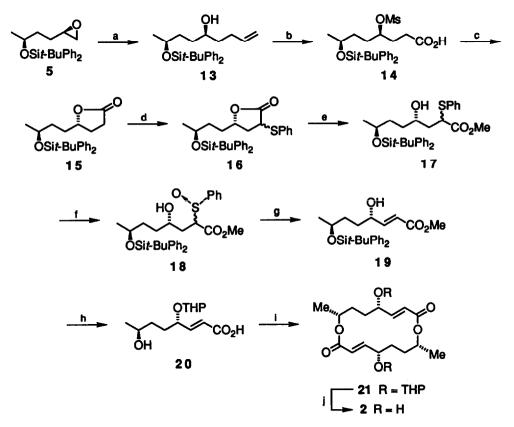
(a) Vitride (1 mol equiv), THF, 0 °C  $\rightarrow$  rt; (b) *t*-BuPh<sub>2</sub>SiCl, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, rt; (c) PhSCH<sub>2</sub>CO<sub>2</sub>H, LDA (2 equiv), THF, 0 °C  $\rightarrow$  rt, then CH<sub>2</sub>N<sub>2</sub>, Et<sub>2</sub>O; (d) NaIO<sub>4</sub>, MeOH-H<sub>2</sub>O, rt; (e) Py (2 equiv), toluene, reflux; (f) PDC, DMF, rt; (g) CH(OEt)<sub>3</sub>, HO(CH<sub>2</sub>)<sub>2</sub>OH, BF<sub>3</sub>•OEt<sub>2</sub>, benzene, reflux; (h) 20% NaOH-MeOH, rt, then Bu<sub>4</sub>NF, THF, reflux; (i) Ph<sub>3</sub>P, DEAD, toluene-THF (10:1), -25 °C, 10 h; (j) CSA, acetone.

oxidation of 6 in aqueous methanol, underwent pyrolysis in boiling toluene in the presence of pyridine to furnish (E)- $\alpha$ , $\beta$ -unsaturated hydroxy ester 8 (85% yield from 6). Compound 8 was converted to 10 (67% yield) via PDC oxidation and subsequent protection of the resulting ketone as the ethylene ketal. Alkaline hydrolysis of the ester followed by desilylation (Bu<sub>4</sub>NF, THF, reflux) provided the hydroxy acid 11 (100%). Cyclodimerization to 12 was carried out under the Mitsunobu conditions according to Gerlach's procedure.<sup>3b</sup> Thus, a reasonably dilute solution of 11 (10<sup>-2</sup> M) in toluene-THF (10:1) was treated with Ph<sub>3</sub>P and DEAD at -25 °C for 10 h to generate the protected pyrenophorin 12 (44% yield). Acidic removal of the carbonyl protecting group furnished (-)-pyrenophorin (1) [mp 177-178 °C (lit.<sup>1</sup> mp 175 °C); [ $\alpha$ ]<sup>25</sup>D -55.9° (c 0.44, EtOH) [lit.<sup>1</sup> [ $\alpha$ ]D -50° (EtOH)<sup>10</sup>], [ $\alpha$ ]<sup>25</sup>D -71.5° (c 0.13, acetone) [lit.<sup>5a</sup> [ $\alpha$ ]D -54.5° (c 0.48, acetone), lit.<sup>6</sup>

 $[\alpha]^{26}_{D}$  -72.9° (c 0.650, acetone)] in 80% yield, whose spectra were in accord with those for natural (<sup>1</sup>H NMR) and synthetic (IR and <sup>13</sup>C NMR)<sup>3b,5b</sup> materials of 1.

We next undertook to elaborate (-)-pyrenophorol (2) by utilizing the foregoing monoepoxide 5 as outlined in Scheme 2. Addition of the Grignard reagent (CH<sub>2</sub>=CHCH<sub>2</sub>MgCl) in the presence of Cul to 5 led to ring opening to give the unsaturated alcohol 13 (76%). O-Mesylation and oxidation with RuO<sub>4</sub> gave the carboxylic acid 14 (82%). Treatment of 14 with KHCO<sub>3</sub> in aqueous methanol resulted in cyclization to give the  $\gamma$ -lactone 15 (90%) as a single diastereomer: Thus, the stereochemistry at C-4 (in 14) was inverted in this S<sub>N</sub>2 reaction. Introduction of a double bond  $\alpha,\beta$  to the carbonyl group was effectively achieved by application of the sulphenylation-dehydrosulphenylation sequence in a similar manner as described above. Accordingly, the enolate of the  $\gamma$ -lactone 15 generated by using LiN(SiMe<sub>3</sub>)<sub>2</sub> (HMPA-THF, -78 °C) was treated with

Scheme 2



(a) CH<sub>2</sub>=CHCH<sub>2</sub>MgCl, CuI, THF, -15 °C; (b) MsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, then RuO<sub>4</sub>, CCl<sub>4</sub>-MeCN-H<sub>2</sub>O, rt; (c) KHCO<sub>3</sub>, MeOH-H<sub>2</sub>O, rt, 10 min; (d) 1 M LiN(SiMe<sub>3</sub>)<sub>2</sub> in THF, THF-HMPA, -78 °C, then PhSSPh, THF, -78 °C; (e) 20% aq NaOH, MeOH, rt, then CH<sub>2</sub>N<sub>2</sub>, Et<sub>2</sub>O, 0 °C; (f) mCPBA, -20 °C; (g) Py (2 equiv), toluene, reflux; (h) i, 2,3-dihydropyran, CSA, CH<sub>2</sub>Cl<sub>2</sub>; ii, 20% aq NaOH, MeOH, rt; iii, Bu<sub>4</sub>NF, THF, reflux; (i) Ph<sub>3</sub>P, DEAD, toluene-THF (10:1), -25 °C, 10 h; (j) TsOH-H<sub>2</sub>O, MeOH, rt.

diphenyl disulfide to afford the  $\alpha$ -phenylthio- $\gamma$ -lactone 16 (72% yield), which was then subjected to hydrolysis followed by diazomethane esterification to give the  $\alpha$ -phenylthio- $\gamma$ -hydroxyester 17 (81% yield). After conversion to the sulfoxide 18 via oxidation with mCPBA, pyrolysis occurred in boiling toluene to generate the (*E*)- $\alpha$ , $\beta$ -unsaturated hydroxy ester 19 (89% from 17). After protected as the tetrahydropyranyl ether, the ester was hydrolyzed, and subsequent desilylation afforded the hydroxy carboxylic acid 20. When 20 was exposed to the Mitsunobu conditions described above for the formation of 12, the macrolactonization took place with complete inversion of chirality at C-4 to furnish 21 in 59% yield. Removal of the O-protecting group (TsOH, MeOH) provided (-)-pyrenophorol (2), mp 138–139 °C (lit.<sup>2a</sup> mp 135 °C); [ $\alpha$ ]<sup>24</sup>D –3.6° (*c* 0.36, acetone) [lit.<sup>2a</sup> [ $\alpha$ ]<sup>20</sup>D –3° (*c* 1.0, acetone)] in 84% yield. The IR (CH<sub>2</sub>Cl<sub>2</sub>), <sup>1</sup>H NMR, and CIMS (isobutane) of synthetic 2 were all identical with those of an authentic specimen of (-)-pyrenophorol.

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## **References and Notes**

- 1. Nozoe, S.; Hirai, K.; Tsuda, K.; Ishibashi, K.; Shirasaka, M. Tetrahedron Lett. 1965, 4675.
- (a) Kis, Z.; Furger, P.; Sigg, H. P. Experientia 1969, 25, 123. (b) Grove, J. F. J. Chem. Soc., C 1971, 2261.
- For syntheses of (±)-pyrenophorin, see: (a) Colvin, E. W.; Purcell, T. A.; Raphael, R. A. J. Chem. Soc., Chem. Commun. 1972, 1031. Colvin, E. W.; Purcell, T. A.; Raphael, R. A. J. Chem. Soc., Perkin Trans. 1 1976, 1718. (b) Gerlach, H.; Oertle, K.; Thalmann, A. Helv. Chim. Acta 1977, 60, 2860. (c) Bakuzis, P.; Bakuzis, M. L. F.; Weingartner, T. F. Tetrahedron Lett. 1978, 2371. (d) Asaoka, M.; Yanagida, N.; Sugimura, N.; Takei, H. Bull. Chem. Soc. Jpn. 1980, 53, 1061. (e) Asaoka, M.; Mukuta, T.; Takei, H. Tetrahedron Lett. 1981, 22, 735. (f) Hase, T. A.; Ourila, A.; Holmberg, C. J. Org. Chem. 1981, 46, 3137. (g) Fujisawa, T.; Takeuchi, M.; Sato, T. Chem. Lett. 1982, 1795. (h) Steliou, K.; Poupart, M.-A. J. Am. Chem. Soc. 1983, 105, 7130. (i) Wakamatsu, T.; Yamada, S.; Ozaki, Y.; Ban, Y. Tetrahedron Lett. 1985, 26, 1989.
- For syntheses of monomeric precursors to pyrenophorin, see: (a) Trost, B. M.; Gowland, F. W. J. Org. Chem. 1979, 44, 3448. (b) Tsuji, J.; Sakai, K.; Nagashima, H.; Shimizu, I. Tetrahedron Lett. 1981, 22, 131. (c) Baraldi, P. G.; Barco, A.; Benetti, S.; Moroder, F.; Pollini, G. P.; Simoni, D. J. Org. Chem. 1983, 48, 1297. (d) Yokota, S.; Nishida, M.; Mitsunobu, O. Bull Chem. Soc. Jpn. 1983, 56, 1803. (e) Labadie, J. W.; Stille, J. K. Tetrahedron Lett. 1983, 24, 4283. (f) Dumont, W.; Vermeyen, C.; Krief, A. *ibid.* 1984, 25, 2883. (g) Derguini, F.; Linstrumelle, G. *ibid.* 1984, 25, 5763. (h) Ngooi, T. K.; Scilimati, A.; Guo, Z.; Sih, C. J. J. Org. Chem. 1989, 54, 911.
- (a) Seebach, D.; Seuring B.; Kalinowski, H.-O.; Lubosch, W.; Renger, B. Angew. Chem. 1977, 89, 270; Angew. Chem., Int. Ed. Engl. 1977, 16, 264. Seuring, B. Seebach, D. Liebigs Ann. Chem. 1978, 2044.
  (b) Mali, R. S.; Pohmakotr, M.; Weidmann, B.; Seebach, D. *ibid.* 1981, 2272.
- 6. Hatakeyama, S.; Satoh, K.; Sakurai, K.; Takano, S. Tetrahedron Lett. 1987, 28, 2717.
- 7. Dommerholt, F. J.; Thijs, L.; Zwanenburg, B. Tetrahedron Lett. 1991, 32, 1499.
- Machinaga, N.; Kibayashi, C. Tetrahedron Lett. 1990, 31, 3637. Machinaga, N.; Kibayashi, C. J. Org. Chem. 1991, 56, 1386. Machinaga, N.; Kibayashi, C. J. Chem. Soc., Chem. Commun. 1991, 405. Machinaga, N.; Kibayashi, C. J. Org. Chem. 1992, 57, 5178.
- 9. For improved synthesis of 3, see: Machinaga, N.; Kibayashi, C. Synthesis in press.
- 10. The reported data in ref 1 for optical rotation of 1 omits description of the solvent (EtOH) used; we obtained this infromation directly upon inquiry of Dr. K. Hirai, one of the authors.

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